#### Monatshefte für Chemie Chemical Monthly

© by Springer-Verlag 1985

# Pyrimido[4,5—*e*] (1,2,4)-triazolo[3,4—*b*] (1,3,4)-thiadiazine-7,9(6*H*,8*H*)-diones

H. K. Gakhar\* and J. K. Gill

Department of Chemistry, Panjab University, Chandigarh-160014, India

(Received 25 April 1984, Accepted 22 May 1984)

Some 5H-pyrimido[4,5—e](1,2,4)-triazolo[3,4—b](1,3,4)-thiadiazine-7,9-(6H,8H)-diones (**4 a**–**d**) have been synthesised by the condensation of 3-alkyl-4-amino-5-mercapto-(1,2,4)-triazoles (**1 a**–**d**) with 5-bromobarbituric acid (**2 a**). Similarly some 9a-nitro-5H-pyrimido[4,5—e](1,2,4)-triazolo[3,4—b](1,3,4)-thiadiazine-7,9(8H,9aH)-diones (**5 a**–**d**) have been obtained by the condensation of **1 a**–**d** with 5-bromo-5-nitrobarbituric acid (**2 b**) and final cyclisation with PPA. The structures have been confirmed by PMR spectra and analytical results.

(Keywords: Heterocyclic compounds)

*Pyrimido*[4,5—e](1,2,4)-triazolo[3,4—b](1,3,4)-thiadiazin-7,9(6H,8H)-dione

Es wurden einige 5H-pyrimido[4,5—e](1,2,4)-triazolo[3,4—b](1,3,4)-thiadiazin-7,9(6H,8H)-dione (4 a—d) mittels Kondensation von 3-Alkyl-4-amino-5-mercapto-(1,2,4)-triazolen (1 a—d) mit 5-Brombarbitursäure (2 a) darge-stellt. Des weiteren wurden einige 9a-Nitro-5H-pyrimido[4,5—e](1,2,4)-triazolo-[3,4—b](1,3,4)-thiadiazin-7,9(8H,9aH)-dione (5 a—d) über die Kondensation von 1 a—d mit 5-Brom-5-nitrobarbitursäure (2 b) und anschließender Cyclisierung mit PPA synthetisiert. Die angeführten Strukturen wurden mittels PMR-Spektren und analytischen Daten abgesichert.

#### Introduction

Some triazolothiadiazines have been reported  $^{1-7}$  to possess antibacterial and antifungal activity. Therefore it seemed to be of interest to introduce another pyrimidine ring. We now report the synthesis of another new system, namely 5H-pyrimido [4,5-e](1,2,4)-triazolo-[3,4-b](1,3,4)-thiadiazines.

#### Results and Discussion

We started with 3-alkyl-4-amino-5-mercapto(1,2,4)-triazoles (1) which in turn were accessible by the treatment of hydrazine hydrate with carbondisulphide followed by reaction with various carboxylic acids.

The 3-alkyl-4-amino-5-mercapto(1,2,4)-triazoles (1 a—d) thus formed were reacted with 5-bromobarbituric acid (2 a) to give the required 3-alkyl-5*H*-pyrimido[4,5—e](1,2,4)-triazolo[3,4—b](1,3,4)-thiadiazine-7,9(6*H*,8*H*)-diones (4 a—d) via the intermediate 3 a'—d' (R=H) which could not be isolated. Structures 4 a—d are supported by analytical results, by PMR spectra and their insolubility in sodium hydroxide solution.

In its PMR-spectrum (chemical shifts in  $\delta$ /ppm) in CDCl<sub>3</sub> + TFA 4c exhibited a quartet (2H) at 3.17 and a triplet (3H) at 1.50 assignable to two and three protons respectively of the ethyl group at position 3. The signal for the proton at 9a did not appear excluding therefore the possibility of structures 5 or 6.

Similarly  $(1 \, a-d)$  on condensation with 5-bromo-5-nitrobarbituric acid  $(2 \, b)$  gave rise to  $3 \, a-d \, (R'=NO_2)$  which on cyclodehydration with PPA could give 5 or 6. These two structures differ only in the position of the C=N double bond. However, we have assigned structure 5 to the cyclised product on the basis of its greater stability because of the conjugation of the carbonyl group with the -C=N bond.

In its PMR-spectrum (CDCl<sub>3</sub> + TFA) **5 c** displayed a quartet (2H) at 3.34 and a triplet (3H) at 1.48 assignable to two and three protons, respectively of the ethyl group at position 3.

Compound 5c was methylated with  $CH_3I$  in the presence of  $K_2CO_3$  to give 7. The PMR spectrum of 7 could not be recorded because of its insolubility.

### Acknowledgements

The authors wish to thank Prof. S. V. Kessar, Chairman, Chemistry Department, for the necessary facilities and to the CSIR New Delhi for the award of junior research fellowship to one of us (J.K.G.).

### **Experimental**

Melting points were determined in open glass capillaries using liquid paraffin bath and are uncorrected. IR spectra were recorded in nujol on a Perkin Elmer 337 and PMR on a Varian EM 390 90 MHz spectrometer using *TMS* as the internal reference. The analytical values (C, H, N) agree with the proposed structures 3 a—d. 4 a—d. and 5 a—d.

3-Ethyl-5H-pyrimido[4,5—e](1,2,4)-triazolo[3,4—b](1,3,4)-thiadiazine-7,9(6H,8H)-dione (
$$\mathbf{4c}$$
,  $R=C_2H_5$ )

A solution of 4-amino-3-ethyl-5-mercepto(1,2,4)-triazole (1)<sup>8,9</sup> (1.44 g, 0.01 mol) in ethanol (40 ml) was added to an ethanolic solution (300 ml) of 5-bromobarbituric acid (2 a) (2.07 g, 0.01 mol) dropwise with continuous stirring at room temperature. There was immediate separation of a new solid which increased as the addition progressed. The contents were stirred at room temperature for 2 h and then refluxed on a steam bath for 0.5 h. After cooling, the solid product was collected under suction, washed with a dilute solution of sodium carbonate and finally with water. Recrystallisation from glacial acetic acid gave colourless needles.

Other 3-alkyl-4-amino-5-mercepto(1,2,4)-triazoles were also condensed with 5-bromobarbituric acid under identical conditions. The data regarding these compounds are collected in Table 1.

Table 1. $3$ -alkyl-5 $H$ -pyrimido[4,5— $e$ ] (1,2,4)-triazolo[3,4— $b$ ] (1.3.4)-thiadia	ızine-
7,9 (6 H,8 H)-diones <b>4 a-d</b>	*

No.	R =	Yield %	M.P.* °C	Molecular Formula
4 a	Н	85	257	$C_6H_6N_6O_3S$
4 b	$CH_3$	90	254	$C_7H_8N_6O_3S$
4 c	$C_2H_5$	93	235	$C_8H_{10}N_6O_3S$
4 d	$n$ - $C_3H_7$	60	242	$C_9H_{12}N_6O_3S$

<sup>\* 4</sup>a and 4b were crystallised from ethanol while 4c and 4d were crystallised from glacial acetic acid. All compounds crystallise with one molecule of water.

5-[ (4-Amino-5-ethyl-4H-1,2,4-triazol-3-yl)thio]-barbituric acid (3 
$$c$$
,  $R=C_2H_5$ ,  $R'=NO_2$ )

To a solution of 5-bromo-5-nitrobarbituric acid (2, 2.52 g, 0.01 mol) in absolute ethanol (300 ml) was added an ethanolic solution (40 ml) of 4-amino-3-ethyl-5-mercepto(1,2,4)-triazole (1 c, 1.44 g, 0.01 mol) dropwise with constant stirring at room temperature when the intermediate  $3 \, c$  separated out immediately. It was collected under suction, washed with a diluted solution of sodium carbonate and finally with cold water.

Other 3-alkyl-4-amino-5-mercepto-(1,2,4)-triazoles were condensed with 5-bromo-5-nitrobarbituric acid under identical conditions. Corresponding data are collected in Table 2.

No.	R =	Yield %	M.P.* °C	Molecular Formula
3 a	Н	53	155	$C_6H_7N_7O_6S$
3 b	$CH_3$	48	188 (d)	$C_7H_9N_7O_6S$
3 e	$C_2H_5$	42	260	$C_8H_{11}N_7O_6S$
3 d	$n$ - $C_3H_7$	45	280	$C_9H_{13}N_7O_6S$

Table 2. 5-[(4-Amino-5-alkyl-4H-1,2,4-triazol-3-yl)thio]-barbituric acid 3 a-d

## 3-Ethyl-9a-nitro-5H-pyrimido[4,5—e](1,2,4)-triazolo[3,4—b](1,3,4)-thiadiazine-7,9(8H,9aH)-dione (5c)

Intermediate 3c (1.00 g) was mixed with a freshly prepared solution of polyphosphoric acid [by mixing phosphorous pentoxide (4 g) and orthophosphoric acid (3 ml)]. The contents were heated on an oil bath at 120–130° for 3 h. It was cooled to room temperature and basified with 10% potassium carbonate solution. The solution was concentrated to 10 ml and the solid thus separated was collected under suction. After washing it with ice-cold water (5 ml), it was crystallised from ethanol.

Other intermediates were also cyclised in a similar fashion. For corresponding data see Table 3.

# 5,8-Dimethyl-9a-nitro-5H-pyrimido[4,5—e](1,2,4)-triazolo[3,4—b](1,3,4)-thiadiazine-7,9(8H,9aH)-dione (7)

A mixture of  $5\,c$  (0.315 g, 0.001 ml) methyl iodide (0.170 g, 0.0012 mol) and  $K_2 CO_3$  (0.069 g, 0.0005 mol) in *DMF* (5 ml) was stirred at 120° for 2 h. The contents were evaporated to dryness and the residue treated with water. The product thus separated was crystallised from ethanol m.p.  $> 360^\circ$ .

<sup>\*</sup> All these compounds were crystallised from water and include one molecule of water.

No.	R =	Yield %	M.P.* °C	Molecular formula
5a	Н	53	268	$C_6H_5N_7O_5S$
5 b	$CH_3$	48	> 360	$C_7H_7N_7O_5S$
5 c	$C_2 H_5$	42	> 360	$C_8H_9N_7O_5S$
5 d	$n$ - $C_3H_7$	45	> 360	$C_9H_{11}N_7O_5S$

Table 3. 3-Alkyl-9a-nitro-5H-pyrimido[4,5—<math>e](1,2,4)-triazolo[3,4—<math>b](1,3,4)thiadiazine-7,9(8H,9aH)diones 5 a-d

#### References

- <sup>1</sup> Upadhyaya V. P., Srinivasan V. R., Indian J. Chem. 16B, 737 (1978).
- <sup>2</sup> Bala S., Gupta R. P., Sachdeva M. L., Singh A., Pujari H. K., Indian J. Chem. 16B, 481 (1978).
- <sup>3</sup> Chadha V. K., J. Indian Chem. Soc. **55**, 817 (1978).
- <sup>4</sup> Pascal J. C., Pinhas H., Ger. Offen. 2,818, 395 (16 Nov., 1978), Brit. Appl. 77/18, 448, 03 May (1977) pp. 17; C.A. **90**, 152246 f (1979).

  <sup>5</sup> Chadha V. K., Sharma G. R., J. Indian Chem. Soc. **57**, 1112 (1980).
- <sup>6</sup> Singh S., Yadav L. D. S., Singh H., Bokin Bobai 8, 385 (1980); C.A. 94, 103250 b (1981).
- <sup>7</sup> Rudnicka W., Osmialowska Z., Acta Pol Pharm. 36, 411 (1979).
- <sup>8</sup> Audrieth L. F., Scott E. S., Kippur P. S., J. Org. Chem. 19, 733 (1954).
- <sup>9</sup> Beyer H., Kroger C. F., Liebigs Ann. **637**, 135 (1960).

<sup>\*</sup> All these compounds were crystallised from ethanol and include one molecule of water.